## ent-Kaurane Diterpenes and Further Constituents from Wedelia trilobata

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Two new *ent*-kaurane diterpenes, wedelidins A (8) and B (9), together with eighteen other constituents, including the sesquiterpenoids 1 and 2, *ent*-kaurane diterpenes 3-7, triterpenoids 10 and 11, steroids 12-14, and flavonoids 15-17 as well as benzene derivatives 18-20, were isolated from the aerial parts of *Wedelia trilobata*. The structures of wedelidins A (8) and B (9) were elucidated by extensive spectroscopic analyses (including UV, IR, NMR, and MS). Furthermore, the structures of compounds 2 and 3 were confirmed by X-ray single-crystal diffraction analyses.

**Introduction.** – Wedelia trilobata L. (Asteraceae) is a perennial plant. In traditional herbal medicine, *W. trilobata* has been used for the treatment of fever and malaria in Vietnam [1]. Previous investigations revealed that the secondary metabolites from this plant mainly consisted of terpenoids [1-3], flavonoids [4], and polyacetylenes as well as steroids [5].

As part of our ongoing search for terpenoids and steroids from natural sources [6–8], we investigated the aerial parts of *W. trilobata* from a phytochemical viewpoint, which led to the isolation of the 20 secondary metabolites 1-20, including the two sesquiterpenoids 1 and 2, the seven *ent*-kaurane diterpenes 3-9, the two triterpenoids 10 and 11, the three steroids 12-14, and the three flavonoids 15-17 as well as the three benzene derivatives 18-20. Among them, compounds 8 and 9, named wedelidin A and B, resp., are two new *ent*-kaurane diterpenes. Herein, we will report the isolation and structural elucidation of these isolates. In addition, the <sup>13</sup>C-NMR data of the five known *ent*-kaurane diterpenes 3-7 were completely assigned here by 2D-NMR spectra (<sup>1</sup>H, <sup>1</sup>H-COSY, HSQC, and HMBC), and the structures of compounds 2 and 3 were further confirmed by X-ray single-crystal diffraction analyses.

**Results and Discussion.** – Wedelidin A (8) was obtained as a white powder. The negative-ion-mode ESI-MS showed a quasimolecular ion at m/z 463.1 ( $[M-H]^-$ ), which, combined with analyses of <sup>13</sup>C-NMR data and the DEPT experiment, indicated a molecular formula C<sub>29</sub>H<sub>36</sub>O<sub>5</sub>. This hypothesis was also supported by the quasimolecular ion at m/z 487.2445 ( $[M + Na]^+$ ) in the positive-ion-mode HR-ESI-MS, corresponding to 12 degrees of unsaturation. Its IR spectrum showed absorptions at 3347 (OH), 1708 (C=O), and 1567 and 1470 cm<sup>-1</sup> (aromatic C=C). The existence of a

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cinnamate (=(2E)-3-phenylprop-2-enoate) ester function was supported by the following <sup>1</sup>H- and <sup>13</sup>C-NMR data:  $\delta(H)$  6.47 (d, J = 15.9 Hz, 1 H), 7.70 (d, J =15.9 Hz, 1 H), 7.37–7.39 (overlap, 3 H), and 7.52–7.54 (overlap, 2 H), and  $\delta(C)$ 166.7 (s), 118.3 (d), 145.1 (d), 134.7 (s), 128.1 (d), 128.9 (d), and 130.3 (d) (Tables 1 and 2). Besides the above moiety, the remaining twenty C-atoms included a trisubstituted C=C bond ( $\delta$ (H) 5.53 (br. s);  $\delta$ (C) 137.1 (d) and 141.3 (s)), a COOH group ( $\delta$ (C) 180.4 (s)), an O-bearing CH group ( $\delta$ (H) 4.70 (dd, J = 4.2, 12.0 Hz);  $\delta$ (C) 78.8 (d)), and an O-bearing CH<sub>2</sub> group ( $\delta$ (H) 4.57 (br. s);  $\delta$ (C) 75.1 (t)). Further analyses indicated a closely similar NMR pattern of compound 8 to that of  $(3\alpha)$ -3-(cinnamoyloxy)-ent-kaur-16-en-19-oic acid (7), suggesting the structure of an *ent*-kaurane diterpenoid for 8 [2]. The cinnamate ester moiety should be bonded to the position C(3) based on the HMBC from H–C(3) to C(1') (Fig. 1). The major difference in the NMR data between compounds 8 and 7 was that the signals of the above-described O-bearing  $CH_2$  group and the trisubstituted C=C bond in ring D of 8 appeared instead of those of the exocyclic  $CH_2 = C(16)$  moiety of 7. The NMR data of ring D of 8 were also supported by the HMBC experiment (Fig. 1). The relative configuration of compound 8 was established by comparison of its NMR data with those of 7. The H–C(3) signal of 8 appeared as dd (J = 4.2 and 12.0 Hz) with a closely similar coupling pattern to that observed for 7 (dd, J = 4.4 and 12.0 Hz) [2]. So, the cinnamate ester moiety should be  $\alpha$ oriented in compound 8. Finally, the structure of wedelidin A (8) was elucidated as  $(3\alpha)$ -3-(cinnamoyloxy)-17-hydroxy-*ent*-kaur-15-en-19-oic acid.

Wedelidin B (9) was obtained as a white powder. Its positive-ion-mode HR-ESI-MS revealed a quasimolecular ion at m/z 463.2483 ( $[M + H]^+$ ), suggesting a molecular

	<b>8</b> <sup>a</sup> )	<b>9</b> <sup>b</sup> )		
CH <sub>2</sub> (1)	2.03 - 2.05(m), 0.99 - 1.12(m)	2.03 - 2.06 (m), 1.10 - 1.12 (m)		
$CH_2(2)$	2.45 $(q, J = 12.0)$ , 1.81 (overlap)	2.45 $(q, J = 12.0)$ , 1.80 (overlap)		
H-C(3)	$4.70 \ (dd, J = 4.2, 12.0)$	4.70 (dd, J = 4.2, 12.0)		
H-C(5)	1.17 (overlap)	1.15 (overlap)		
$CH_2(6)$	$1.88 - 1.91 \ (m), \ 1.70 - 1.73 \ (m)$	1.88 - 1.92 (m), 1.71 - 1.75 (m)		
$CH_2(7)$	1.93 - 1.96 (m), 1.09 - 1.11 (m)	1.99-2.02(m), 1.14-1.18(m)		
H–C(9)	1.21 - 1.24 (m)	1.24 - 1.29(m)		
CH <sub>2</sub> (11)	1.55-1.60 (overlap, 2 H)	1.55-1.60 (overlap, 2 H)		
$CH_2(12)$	1.58 (overlap), 1.50 (overlap)	1.60 (overlap), 1.50 (overlap)		
H-C(13)	2.62 (br. s)	3.06 (br. s)		
CH <sub>2</sub> (14)	1.89 - 1.91 (m), 1.50 - 1.54 (m)	2.06 - 2.09(m), 1.47 - 1.51(m)		
H-C(15)	5.53 (br. s)	6.59 (br. s)		
CH <sub>2</sub> (17) or H–C(17)	4.57 (br. s)	9.75 (s)		
Me(18)	1.35(s)	1.34(s)		
Me(20)	1.09(s)	1.09(s)		
H–C(2')	6.47 $(d, J = 15.9)$	6.47 (d, J = 16.2)		
H–C(3')	7.70 (d, J = 15.9)	7.70 (d, J = 16.2)		
H <sub>o</sub>	7.52-7.54 (overlap, 2 H)	7.52-7.54 (overlap, 2 H)		
H <sub>m</sub>	7.37–7.39 (overlap, 2 H)	7.37 – 7.39 (overlap, 2 H)		
$H_p$	7.37–7.39 (overlap)	7.37–7.39 (overlap)		
<sup>a</sup> ) Measured at 300 MHz.	<sup>b</sup> ) Measured at 600 MHz.			

Table 1. <sup>1</sup>H-NMR Data (CDCl<sub>3</sub>) of Compounds 8 and 9. δ in ppm, J in Hz.

	<b>3</b> <sup>a</sup> )	<b>4</b> <sup>a</sup> )	<b>5</b> <sup>a</sup> )	<b>6</b> <sup>a</sup> )	<b>7</b> <sup>a</sup> )	<b>8</b> <sup>b</sup> )	9°)
C(1)	41.1 ( <i>t</i> )	40.7 (t)	38.8 (t)	38.8 (t)	38.9 (t)	38.8 (t)	39.7 (t)
C(2)	19.5(t)	20.1(t)	24.2(t)	24.0(t)	24.2(t)	24.1(t)	24.1(t)
C(3)	38.2(t)	38.2(t)	78.7(d)	78.8(d)	79.0 (d)	78.8(d)	78.6(d)
C(4)	44.7(s)	44.7 (s)	48.0 (s)	48.1 (s)	48.1 (s)	48.1 (s)	48.0(s)
C(5)	57.5 (d)	46.6(d)	56.5(d)	56.4(d)	56.5(d)	56.1(d)	55.9 (d)
C(6)	22.3(t)	18.4(t)	21.5(t)	21.5(t)	21.6(t)	20.5(t)	20.1(t)
C(7)	41.7(t)	29.7(t)	39.5 (t)	39.5 (t)	39.5(t)	39.1(t)	38.0(t)
C(8)	44.2(s)	42.3(s)	43.8 (s)	43.8 (s)	43.9 (s)	49.0 (s)	50.6 (s)
C(9)	55.6 (d)	158.5 (s)	55.2 (d)	55.1 (d)	55.2 (d)	47.4(d)	46.1 ( <i>d</i> )
C(10)	40.1 (s)	38.8 (s)	43.8 (s)	43.8 (s)	39.4 (s)	39.6 (s)	38.8(s)
C(11)	18.4(t)	114.9 (d)	18.5(t)	18.5(t)	18.5(t)	19.0(t)	18.8(t)
C(12)	33.5(t)	37.9(t)	33.1(t)	33.0(t)	33.1(t)	25.3(t)	25.1(t)
C(13)	44.3 (d)	41.2(d)	43.9(d)	43.9 (d)	43.8(d)	41.5(d)	37.8 (d)
C(14)	40.1(t)	44.9 (t)	41.0(t)	41.0(t)	41.0(t)	43.6 (t)	42.8(t)
C(15)	49.4(t)	50.3(t)	48.8(t)	48.7(t)	48.8(t)	137.1(d)	161.0(d)
C(16)	156.3 (s)	155.9 (s)	155.3 (s)	155.3 (s)	155.3 (s)	141.3 (s)	148.8(s)
C(17)	103.7(t)	105.5(t)	103.3(t)	103.3(t)	103.3(t)	75.1(t)	189.4(d)
C(18)	29.4(q)	28.2(q)	24.0(q)	23.9(q)	23.9(q)	23.8(q)	23.8(q)
C(19)	184.0(s)	183.3 (s)	180.7(s)	180.4(s)	180.6(s)	180.4(s)	180.3(s)
C(20)	16.0(q)	23.6(q)	15.4(q)	15.3(q)	15.4(q)	15.4(q)	15.5(q)
C(1')			167.7 (s)	167.7 (s)	166.8(s)	166.7(s)	166.7(s)
C(2')			128.0(s)	128.8(s)	118.4(d)	118.3(d)	118.2(d)
C(3')			138.0(d)	137.4 (d)	145.1(d)	145.1 (d)	145.2(d)
$Me(4')$ or $C_{inso}$			15.7(q)	14.4(q)	134.5(s)	134.7 (s)	134.4(s)
$Me-C(2')$ or $C_o$			20.6(q)	12.0(q)	128.1(d)	128.1(d)	128.1(d)
C <sub>m</sub>			.17	.17	128.8(d)	128.9 (d)	128.9 (d)
$C_p$					130.3 ( <i>d</i> )	130.3 ( <i>d</i> )	130.3 ( <i>d</i> )

Table 2. <sup>13</sup>C-NMR Data (CDCl<sub>3</sub>) of Compounds 3-9.  $\delta$  in ppm.

 $^{\rm a})$  Measured at 100 MHz.  $^{\rm b})$  Measured at 75 MHz.  $^{\rm c})$  Measured at 150 MHz.



Fig. 1. Key HMBCs  $(H \rightarrow C)$  of compound 8

formula  $C_{29}H_{34}O_5$ , which indicated 13 degrees of unsaturation. The IR spectrum showed absorptions at 3375 (OH), 1710 (C=O), and 1662 and 1453 cm<sup>-1</sup> (aromatic C=C). The presence of a cinnamate ester moiety was supported by NMR data (*Tables 1* and 2). Besides this ester moiety, the remaining twenty C-atoms included a trisubstituted C=C bond ( $\delta$ (H) 6.59 (br. *s*);  $\delta$ (C) 161.0 (*d*) and 148.8 (*s*)), a COOH group ( $\delta$ (C) 180.3 (*s*)), an aldehyde C=O group ( $\delta$ (C) 189.4 (*d*)), and an O-bearing CH group ( $\delta$ (H) 4.70 (*dd*, *J* = 4.8, 12.0 Hz);  $\delta$ (C) 78.6 (*d*)). Further analyses demonstrated

that compound **9** showed a closely similar NMR pattern to that of **8**, besides the existence of a CHO instead of an O-bearing CH<sub>2</sub> group, indicating that compound **9** was an *ent*-kaurane diterpenoid. Based on the HMBCs C(17) ( $\delta$ (C) 189.4 (*s*))/H–C(13) and H–C(15), the position of the CHO group should be assigned to CH(17). The *a*-orientation of the cinnamate ester moiety of **9** was established by comparison of the H–C(3) coupling pattern (*dd*, *J*=4.8 and 12.0 Hz) with that of **8** (*dd*, *J*=4.2 and 12.0 Hz). Thus, the structure of wedelidin B (**9**) was determined as (3 $\alpha$ )-3-(cinnamoyloxy)-17-oxo-*ent*-kaur-15-en-19-oic acid.

In addition to the above described two new ent-kaurane diterpenes 8 and 9, 18 secondary metabolites were isolated. Based on the spectroscopic analyses and comparison with the literature data, they were determined as ivalin (1) [9], wedeliatrilolactone B (2) [10], ent-kaur-16-en-19-oic acid (3) [11], ent-kaura-9(11),16dien-19-oic acid (4) [12],  $(3\alpha)$ -3-(angeloyloxy)-ent-kaur-16-en-19-oic acid (5) [13],  $(3\alpha)$ -3-(tiglinovloxy)-ent-kaur-16-en-19-oic acid (6) [2],  $(3\alpha)$ -3-(cinnamovloxy)-entkaur-16-en-19-oic acid (7) [2],  $\beta$ -friedelinol (10) [14], friedelin (11) [15], stigmasterol (12) [16],  $(7\alpha)$ -7-hydroxystigmasterol (13) [17],  $(3\beta)$ -3-hydroxystigmasta-5,22-dien-7one (14) [18], 3-hydroxy-6-methoxychromen-4-one (15) [19], apigenin (16) [20], diosmetin (17) [21], benzeneacetic acid 2-phenylethenyl ester (18) [22], isocinnamic acid (19) [23], and 4-methoxycatechol (20) [24]. Among these isolates, compounds 1, 4, and 11-20 are reported as secondary metabolites from W. trilobata for the first time. The structures of compounds 2 and 3 were further confirmed by an X-ray single-crystal diffraction experiment for the first time (Fig. 2). Crystals of 2 and 3 were obtained from a solution of CHCl<sub>3</sub>/MeOH 1:1. The crystallographic data have been deposited with the *Cambridge Crystallographic Data Centre*<sup>1</sup>).



Fig. 2. X-Ray single-crystal structures of compounds 2 and 3. Arbitrary atom numbering.

CCDC-778708 (for 2) and 782281 (for 3) contain the supplementary crystallographic data for this article. These data can be obtained free of charge from the *Cambridge Crystallographic Data Centre via* www.ccdc.cam.ac.uk/data\_request/cif.

This work was supported by the *Basic Research Program* (973 Program) of China (No. 2007CB108903), the National Science Foundation of China (No. 20972062 and No. 30970556), and the 111 Project.

## **Experimental Part**

General. Column chromatography (CC): silica gel (SiO<sub>2</sub>, 200–300 mesh; Qingdao Marine Chemical Factory, Qingdao, P. R. China). TLC: SiO<sub>2</sub> GF<sub>254</sub> (10–40 µm, Qingdao Marine Chemical Factory); detection at 254 nm, and by heating after spraying with 98% H<sub>2</sub>SO<sub>4</sub> soln./EtOH 5:95 ( $\nu/\nu$ ). Optical rotations: Perkin-Elmer-341 polarimeter; in MeOH at 25°. UV Spectra: NewCentury-Pgeneral-T6 spectrophotometer;  $\lambda_{max}$  (log  $\varepsilon$ ) in nm. IR Spectra: Nicolet-Nexus-670 FT-IR spectrometer; with KBr pellets;  $\tilde{\nu}$  in cm<sup>-1</sup>. NMR Spectra: Varian-Inova-300, Bruker-Avance-III-400, and Varian-Inova-600 instruments;  $\delta$  in ppm rel. to Me<sub>4</sub>Si as internal standard, J in Hz. MS: Bruker-Esquire-6000 (ESI) and Bruker-Apex-II (HR-ESI) instrument; in m/z.

*Plant Material.* The plant material was collected in Haikou City, Hainan Province, China, in June 2008 and authenticated by advanced lab assistant *Qiong-Xin Zhong* of the College of Biology, Hainan Formal University. A voucher specimen (20070805) was deposited with the College of Chemistry and Chemical Engineering, Lanzhou University.

Extraction and Isolation. The air-dried whole plant of W. trilobata (4600.3 g) was powdered and extracted with 95% EtOH (151) three times (each for 7 d) at r.t. and the soln. concentrated to give a crude extract (671.2 g), which was suspended in  $H_2O$  (2.5 l) and then extracted with petroleum ether  $(60-90^\circ; 3 \times 2.51)$  and CHCl<sub>3</sub>  $(4 \times 2.01)$  successively. The petroleum ether extract (180.1 g) was subjected to CC (SiO<sub>2</sub>, petroleum ether/AcOEt 40:1, 20:1, 10:1, 5:1, 2:1, and 0:1): Fractions 1-6(monitored by TLC). Fr. 1 (60.7 g) was repeatedly applied to CC (SiO<sub>2</sub>, petroleum ether/AcOEt 50:1 $\rightarrow$ 10:1): 3 (10.8 g) and 4 (68.6 mg). Fr. 2 (10.5 g) was also applied to CC (SiO<sub>2</sub>, petroleum ether/AcOEt 25:1): 5 (2.1 mg) and a crude crystalline product. The latter was recrystallized from petroleum ether/ AcOEt 1:1: 9 (3.8 g). Fr. 3 (20.4 g) was subjected to repeated CC (SiO<sub>2</sub>, petroleum ether/AcOEt  $20:1 \rightarrow 5:1$ ): Frs. 3.1 and 3.2. Fr. 3.1 (10.2 g) was further purified by CC (SiO<sub>2</sub>, petroleum ether/AcOEt 10:1): 6 (8.2 mg), 7 (20.6 mg), and 8 (4.6 mg). Fr. 3.2 (0.6 g) was subjected to prep. TLC (CHCl<sub>3</sub>/AcOEt 15:1): 10 (4.7 mg) and 11 (33.2 mg). Fr. 4 (3.8 g) was applied to CC (SiO<sub>2</sub>, petroleum ether/AcOEt  $10:1 \rightarrow 2:1$ ): Frs. 4.1-4.3. Fr. 4.1 (1.0 g) was further subjected to CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/AcOEt 15:1  $\rightarrow$  5:1): 13 (2.9 mg), 15 (6.8 mg), and 17 (9.2 mg). Fr. 4.2 (1.8 g) was also further purified by CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/ AcOEt  $10:1 \rightarrow 2:1$ ): **1** (1.1 mg), **2** (4.2 mg), and **12** (6.2 mg). Fr. 5 (1.6 g) was subjected to repeated CC  $(SiO_2, CHCl_3/AcOEt 15: 1 \rightarrow 3: 1): 16 (8.3 mg), 19 (10.4 mg), and 20 (5.1 mg). The CHCl_3 extract (5.6 g)$ was subjected to repeated CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/acetone  $50:1 \rightarrow 2:1$ ): Frs. a-d (by TLC). Fr. c (1.9 g) was further purified with CC (SiO<sub>2</sub>, CHCl<sub>3</sub>/CH<sub>3</sub>OH 20:1): 14 (3.0 mg) and 18 (5.6 mg).

Wedelidin A (=( $3\alpha$ )-3-(Cinnamoyloxy)-17-hydroxy-ent-kaur-15-en-19-oic Acid = ( $3\alpha$ )-17-Hydroxy-3-{[(2E)-1-oxo-3-phenylprop-2-en-1-yl]oxy}-ent-kaur-15-en-19-oic Acid; **8**): White amorphous powder. [a]<sub>25</sub><sup>25</sup> = -8.2 (c = 0.1, MeOH). UV (MeOH): 207 (3.25), 274 (3.37). IR (KBr): 3347, 2962, 2931, 2868, 1708, 1567, 1470, 1311, 812. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): Table 1. <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>): Table 2. HR-ESI-MS: 487.2445 ([M + Na]<sup>+</sup>, C<sub>29</sub>H<sub>36</sub>NaO<sup>+</sup><sub>5</sub>; calc. 487.2455).

Wendelidin B (=(3a)-3-(Cinnamoyloxy)-17-oxo-ent-kaur-15-en-19-oic Acid = (3a)-17-Oxo-3-{[(2E)-1-oxo-3-phenylprop-2-en-1-yl]oxy]-ent-kaur-15-en-19-oic Acid; **9**): White amorphous powder. [a]<sub>D</sub><sup>25</sup> = -6.1 (c = 0.1, MeOH). UV (MeOH): 215 (3.11), 275 (3.09). IR (KBr): 3375, 2925, 1710, 1662, 1453, 1070, 1026, 768. <sup>1</sup>H-NMR (600 MHz, CDCl<sub>3</sub>): Table 1. <sup>13</sup>C-NMR (150 MHz, CDCl<sub>3</sub>): Table 2. HR-ESI-MS: 463.2483 ([M + H]<sup>+</sup>, C<sub>29</sub>H<sub>35</sub>O<sup>+</sup><sub>5</sub>; calc. 463.2479).

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Received August 10, 2010